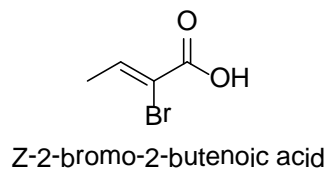
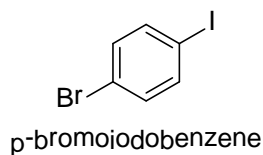
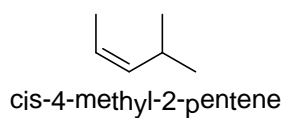
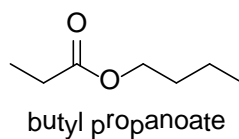
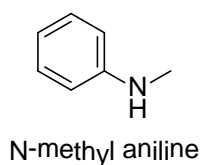
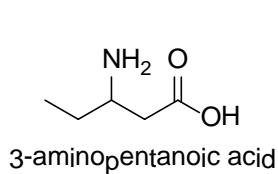


Christmas present 2012 Answers

Problem 1.

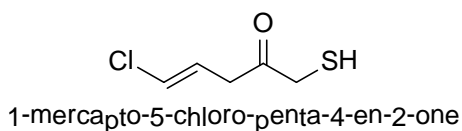


Problem 2.

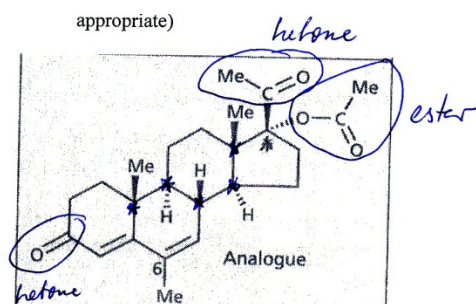
- Indicates halogen
- indicates sulphur in the sample
- shows Cl
- either aldehyde or ketone
- either triple, double bond or activated benzene ring
- Knowing the presence of Cl, S and C=O this leaves $150.5 - 95.5 = 55$ which is C_4H_7 giving a formula C_5H_7ClOS . This means $DBE = 2 \times 5 + 2 - 7 - 1/2 = 2$.

One DBE is going into C=O leaving one as a double bond (not ring) as we can add Br_2 .

One out of many suggestions could be



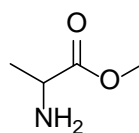
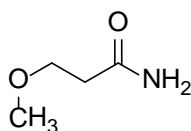
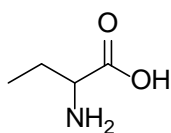
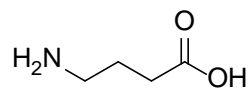
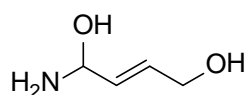
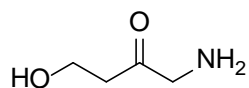
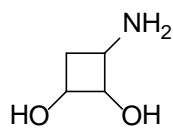
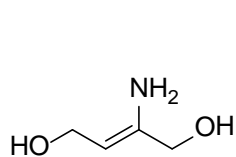
Problem 3.



- Identify the centres of chirality if any are present
- How many stereoisomers are possible?

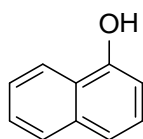
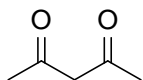
$$2^6 = 64$$

Problem 4.

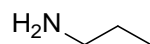
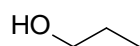
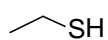


- a) Amine, and alcohol, Z
- b) cyclic
- c) Amine, alcohol and ketone
- d) Trans
- e) Carboxylic acid and amine
- f) Amino acid
- g) Ether and amide
- h) Ester and amine

Problem 5.



Problem 6.



a)

b)

c)